

Isotope-Selective Infrared Spectroscopy Reveals Pathological Changes in the Liver

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BIOGRAPHICAL NOTES

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KEY WORDS

¹³C-methacetin, dose, cumulative dose, infrared spectroscopy, transmittance, absorbance, cirrhosis

ABSTRACT

An overview of new diagnostic device IRIS (Wagner Analysen Technik, Germany) is provided. This device uses the ¹³C stable isotope together with non-dispersive infrared selective spectroscopy for non-invasive quantification of hepatic function parameters, including kinetics (type of liver damage) and capacity (percentage of recovery yield of the liver). The short physical principle of infrared spectroscopy and acousto-optical detector of IRIS was presented. In the final section we verified the diagnostic quality of this new diagnostic device (using ANOVA) on cohort of 76 patients divided into 5 groups according to diagnosis and the various level of liver damage.

INTRODUCTION

Investigation of liver function stretches the entire history of modern hepatology.

Once widespread chromoexcretory tests such as bromsulfophtalein test or so-called galactose elimination test was gradually abandoned and replaced by metabolic tests using radioactive isotopes, which also had their significant limitations. Isotope ^{13}C -labeled substances have the last word in the quantification of liver function. They occur in nature, do not emit ionizing radiation and are therefore completely safe. The principle of testing is simple - proband consumes appropriate material labeled by ^{13}C carbon, for example methacetin. It is metabolized in the liver to carbon dioxide and the amount of $^{13}\text{CO}_2$ in exhaled air is a measure of the intensity of hepatic metabolism. Test results correlate very well with Child-Pugh classification of liver cirrhosis and in addition are applicable for monitoring changes in all liver functions. For the determination of $^{13}\text{CO}_2$ in exhaled air is used infrared spectroscopy isotope. We ourselves use for several years in our department apparatus IRIS (In-fraRed Nondispersive Isotope-Selective Spectroscopy) from Wagner Analysen Technik, Germany [1]. IRIS is supplied with software which is able to distinguish between **normal liver, stimulated hepatic function** (metabolism of the liver is very fast, stimulated by permanent degradation of toxic substances (alcohol, hepatotoxic drugs, etc.), **liver cirrhosis** (metabolism of the liver is highly damaged). The values of measured quantities, i.e. % ^{13}C dose/hr and % ^{13}C cumulated dose in the expired air sample for these three cases, together with reference ranges that define the zones of normal hepatic function, are shown in Fig. 1 and 2.

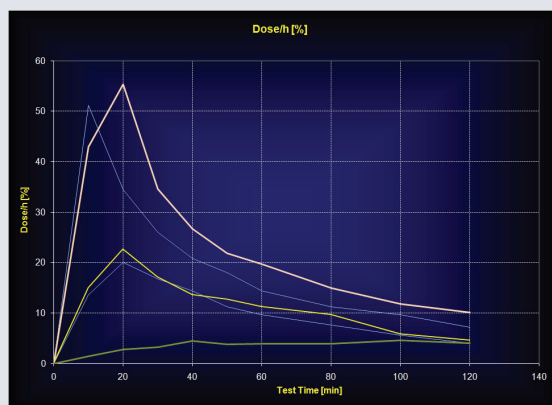


Fig. 1 ^{13}C -methacetin dose/hr [%]. Yellow line - normal liver function, Pink line - stimulated liver function, Green line - cirrhosis; Blue lines indicate the reference ranges of normal values

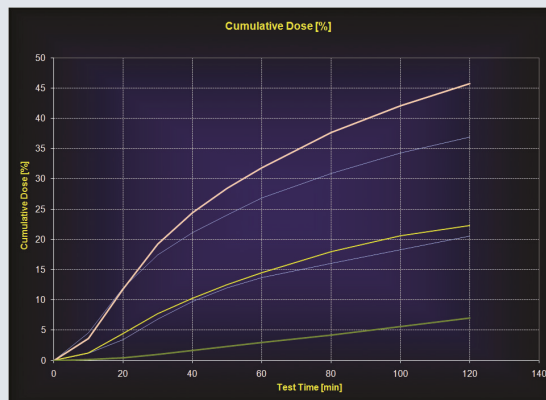


Fig. 2 ^{13}C -methacetin cumulated dose [%]. Yellow line - normal liver function, Pink line - stimulated liver function, Green line - cirrhosis; Blue lines indicate the reference ranges of normal values

MAIN PRINCIPLES

^{13}C -Methacetin

^{13}C commonly occurs in nature, and under normal conditions it represents 1,1% of the total carbon content in the human body. ^{13}C -methacetin is metabolized as well as widespread carbon ^{12}C , it is non-toxic, non-radioactive, and represents a stable isotope, therefore suitable for use in diagnostics.

^{13}C -methacetin is metabolized in the hepatic mixed-function oxidase system using subgroups of the P450 cytochrome (CYP2E1 and CYP1A2) by demethylation/decarboxylation to acetaminophen and CO_2 . The formed carbon dioxide is quickly absorbed into the blood, circulates to the lungs and is exhaled. Use of ^{13}C -marked methacetin allows quantification of hepatic function parameters using the $^{13}\text{CO}_2/^{12}\text{CO}_2$ ratio in exhaled breath; these parameters include kinetics (% ^{13}C dose/hr) that defines the type of liver damage, and capacity (% ^{13}C cumulated dose) that determines the percentage of recovery yield. Additionally, since cytochrome CYP2E1 is active in alcohol metabolism, and is also damaged and stimulated by alcohol, the ^{13}C -methacetin breath test is particularly valuable in assessing hepatic diseases caused by alcohol [1]. The ^{13}C -methacetin breath test is also used to determine the stage of NASH (non-alcoholic steatohepatitis, characterized by hepatic involvement, morphologically similar to hepatic involvement in alcohol abuse) or ASH (alcoholic steatohepatitis - alcoholic liver damage), and assessing liver fibrosis, steatosis, cirrhosis; the test provides information

on the reserve capacity of the liver and is also a very suitable instrument for accurate timing of liver transplantation [2].

INFRARED SPECTROSCOPY

Infrared spectroscopy is an analytical technique designed for identification and structural characterization of organic compounds and inorganic substances. It is based on an interaction of the measured sample with infrared radiation (electromagnetic waves with wavelengths from 800 nm to 1 mm). The energy of infrared photons (1 kJ/mol - 60 kJ/mol) is not sufficient to excite electrons in molecular orbits; however, it is sufficient to change the vibrational or rotational state of molecules. When the change of vibrational and rotational states is associated with a change of dipole moments, absorption of radiation occurs, which is characteristic for a given bond within the molecule. Infrared spectroscopy observes the amount of infrared radiation absorbed as a function of the wavelength. The amount of absorbed radiation is expressed either as percent of transmittance (% T) or in units of absorbance (A). Transmittance is defined as the ratio of intensities of passed radiation (I) to the original beam (I_0) [3]:

$$T = \frac{I}{I_0} \quad (1)$$

Absorbance determines how much radiation was absorbed by the measured sample. Using transmittance, it is defined as follows:

$$A = \log(1/T) = -\log(T) = c \cdot l \cdot \varepsilon \quad (2)$$

where c = molar concentration; l = cell length; ε = molar absorption coefficient, tabulated.

This relationship is also called the Lambert - Beer law; according to this law, absorbance is proportional to the concentration of the absorbing substance [4].

Analytical output of infrared spectroscopy takes the form of infrared spectra, thus a graphical depiction of functional dependence of absorbed energy, expressed precisely by transmittance or absorbance, relative to wavelength. The infrared spectrum is characteristic for individual substances to such an extent that there are virtually no two compounds with an entirely identical infrared

spectrum. An example of a typical infrared spectrum is shown in Fig. 3.

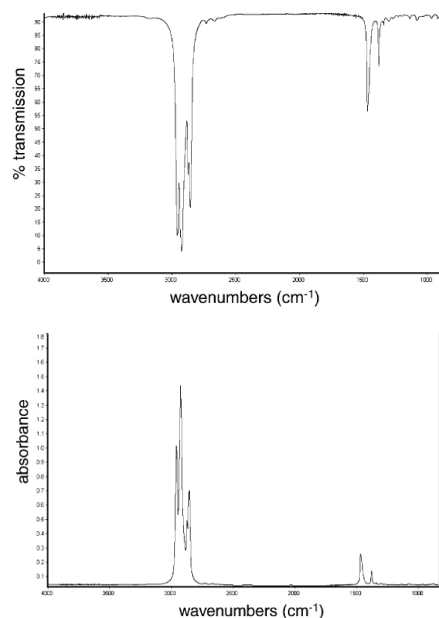


Fig. 3 IR spectrum of the analyzed sample. Transmittance (up), absorbance (down)

Together with non-dispersive infrared spectroscopy, IRIS also uses a broadband light source and an acousto-optical detector (of the Luft - Lehrer type), which is sensitive only to those wavelengths at which the examined gases are IR-absorbing. With two such detectors, which are individually sensitive to the absorption spectra of $^{12}\text{CO}_2$ and $^{13}\text{CO}_2$, the concentrations of these two gas components to be related to each others for $^{13}\text{C}/^{12}\text{C}$ -isotope ratio determination are measured [2]. The IR-absorption spectra of the asymmetrical stretch oscillations modes of both the $^{13}\text{CO}_2$ - and $^{12}\text{CO}_2$ -molecules are separated almost completely.

Luft - Lehrer Detector

The basic scheme of the detector is shown in Fig. 4 [5]. The detector consists of a sealed chamber where molecules of the absorbing substance, which may be present in the sample gas, are placed. The flexible membrane plate of the capacitor partitions the rear section of the detector where the solid plate capacitor is located, from the front, absorbing part, where the window with high optical transmittance is located, and through which the chopped infrared radiation enters the chamber. A detector sig-

nal arises from the change in the capacitor voltage generated by the displacement of the flexible or movable capacitor plate. The gas in the absorption compartment is heated by the IR energy entering through the optical window. With the temperature rise, there is a corresponding pressure increase and the gas expands against the membrane. Chopped IR beams produce membrane displacement oscillations at the chopping frequency. The intensity of the IR beam entering the detector is inversely proportional to the concentration of the absorbing species in the sample cell.

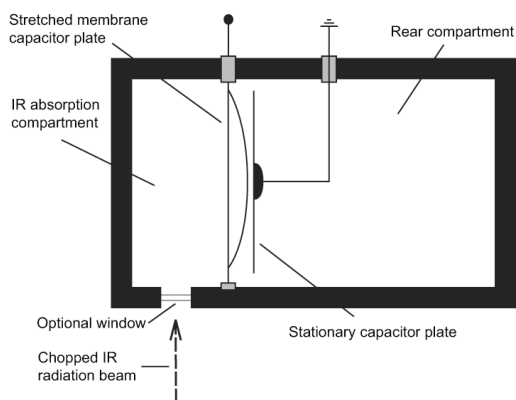


Fig. 4 Acousto-optical detector of the Luft-Lehrer type

SUBJECTS AND METHODS

Patients come in a fasting state for the procedure; they breathe into the first breath bag before consuming the substrate drink (provides baseline ^{13}C -methacetin). Immediately, they consume the test drink (75 mg of ^{13}C -methacetin dissolved in 200 ml

of unsweetened tea or water). Breath samples were collected at 10, 20, 30, 40, 50, 60, 80, 100 and 120 minutes following ingestion of the ^{13}C -methacetin substrate according to the protocol of Liver Function 2 (IRIS User's Guide). Samples were collected by having patients take a deep breath and slowly exhale into additional breath bags.

IRIS is capable to distinguish between a normal, stimulated, or cirrhotic liver. The main goal of our study was to determine whether there were statistically significant differences between the measured values of the dose and cumulated dose of ^{13}C -methacetin measured using IRIS among 5 different patient groups which had been divided according to diagnosis for all time intervals. Additionally, the goal was to determine which sample times show the greatest difference, in other words, which sample time or times provide the "highest diagnostic power". The measurement was done on 76 patients [6], [7]. Group 1 consisted of patients with no serious disease - the control group (N=11); group 2 consisted of patients with cirrhosis (N=10); group 3 consisted of patients with compensated ischemic heart failure (N=30); group 4 consisted of patients with concurrent nephrological and cardiovascular disease (N=10); and group 5 consisted of patients with decompensated heart failure (N=15). Group 4 and 5 were included to determine whether severe liver damage is present in severe heart damage or renal insufficiency.

Analysis of variance (ANOVA) with subsequent application of the Bonferroni's test for multiple comparisons was used to evaluate the data. STATISTICA,

Test Time [min.]	Dose/hr [%]					Cumulated dose [%]				
	1	2	3	4	5	1	2	3	4	5
10	24,82	2,87	22,05	27,40	10,22	2,07	0,24	1,84	2,28	0,85
20	30,89	6,68	28,67	32,22	14,63	6,71	1,03	6,06	7,25	2,92
30	23,51	6,62	24,44	25,41	13,91	11,24	2,14	10,49	12,05	5,30
40	18,65	6,02	20,00	20,94	12,93	14,76	3,20	14,19	15,92	7,54
50	15,68	5,48	16,70	17,64	10,94	17,62	4,15	17,25	19,13	9,53
60	13,45	5,72	14,42	15,54	9,64	20,05	5,09	19,85	21,90	11,24
70	10,90	5,05	11,48	12,71	8,41	24,11	6,88	24,16	26,60	14,25
100	8,88	4,73	9,55	10,57	7,84	27,40	8,51	27,67	30,48	16,69
120	7,70	4,39	8,13	8,86	6,92	30,17	10,03	30,61	33,72	19,42

Tab. 1 Mean values of the dose and the cumulated dose of ^{13}C -methacetin at the times (10, 20, 30, 40, 50, 60, 80, 100, 120 min). 1-Control group; 2-Cirrhosis; 3-Stable heart failure (HF); 4-Concurrent Nephrological and Cardiovascular disease; 5-Decompensated HF

Test Time [min.]	Dose/hr [%]		Cumulated dose [%]	
	F	P	F	P
10	13,374	0,000000	13,374	0,000000
20	15,606	0,000000	15,397	0,000000
30	18,377	0,000000	17,459	0,000000
40	16,380	0,000000	19,209	0,000000
50	14,020	0,000000	20,185	0,000000
60	13,684	0,000000	20,484	0,000000
80	13,240	0,000000	20,308	0,000000
100	9,112	0,000005	19,810	0,000000
120	7,219	0,000062	19,136	0,000000

Tab. 2 ANOVA test results. Time intervals with the highest discrimination ability are highlighted.

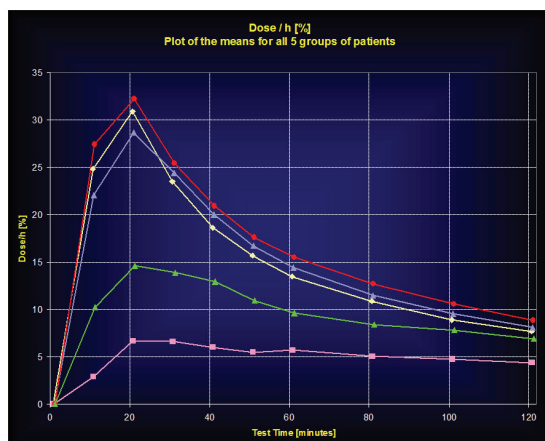


Fig. 5 Graph of mean values and confidence intervals ^{13}C dose/hr [%]. Yellow line-Controls, Pink line-Cirrhosis, Blue line-Compensated Heart Failure, Red line - Concurrent Nephrological and Cardiovascular Disease, Green line-Decompensated Heart Failure

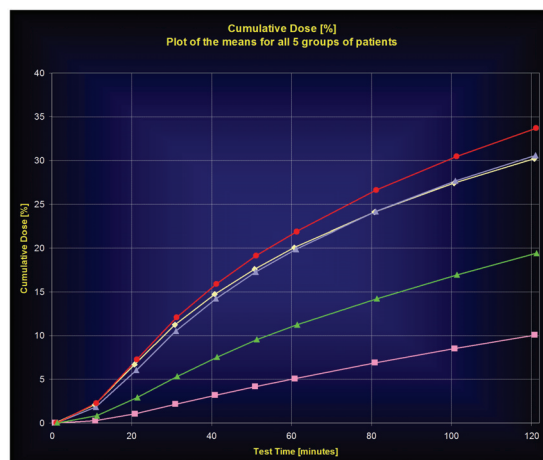


Fig. 6 Graph of mean values and confidence intervals ^{13}C cumulated dose [%]. Yellow line-Controls, Pink line - Cirrhosis, Blue line - Compensated Heart Failure, Red line-Concurrent Nephrological and Cardiovascular Disease, Green line - Decompensated Heart Failure

version 9 from StatSoft Inc., was used for statistical analysis.

RESULTS

Table 1 shows mean values of the measured dose and the cumulated dose of ^{13}C -methacetin in all time intervals for all 5 patient groups. Tab.2 summarizes ANOVA results-only F-test values and values of obtained significance level p.

Table 2 shows that significance levels are very low for all times; $p=0,00$ for almost all sample times. The discrimination power of IRIS can be thus evaluated according to the F-test value; the higher the F value, the lower the significance level, and the higher

the discrimination power. It can be concluded from the values that the highest discrimination power for the ^{13}C dose occurs at 30 and 40min., while the highest discrimination power for the cumulated dose occurs at 50, 60 and 80min. Fig. 5 and 6 present graphs of mean values and confidence intervals for the measured dose value and the cumulated dose of ^{13}C for all 5 patient groups.

CONCLUSIONS

The results show clearly that slightly stimulated hepatic function is present in patients with concurrent nephrological and cardiovascular disease (this is probably due to long-term use of large amounts

of drugs many of which may induce the liver enzymes). In patients with decompensated heart failure, we observed the presence of highly impaired hepatic function, approaching liver cirrhosis. Results of the study are in absolute agreement with those expected and indicate that IRIS is a diagnostic apparatus of superior quality, which has a high ability to effectively distinguish between various degrees of liver damage.

REFERENCES

- [1] Breath testing in clinical Diagnose [online] [cit 2010-06-18], <http://www.wagner-bremen.de/>
- [2] Petrolati A., Festi D., De Berardinis G., Colaiocco-Ferrante L., Di Paolo D., Tisone G., Angelico M.: 13C-Methacetin Breath Test For Monitoring Hepatic Function in Cirrhotic Patients Before and After Liver Transplantation, *Aliment Pharmacol. Ther.* 2004 Jan 15; 19(2), 243
- [3] Infrared Spectroscopy [online], [cit 2010-06-16], Faculty of Education, Charles University in Prague, Chemistry & Chemistry Didactics Dept., downloadable materials: <http://user-web.pedf.cuni.cz/kch/downloads/materialy/INA/ICSpektroskopie.pdf>
- [4] Absorbance [online], [cit 2010-06-18], from Wikipedia, the free encyclopedia, <http://cs.wikipedia.org/wiki/Absorbance>
- [5] Lucero D. P.: An analytical model of the pneumatic nondispersive infrared detector, *J. Phys. E.* 1973; 6, 281-5
- [6] Hendrichova M., Malek F., Kratka K., Sedlakova M., Vranova J., Horak J.: Microsomal liver function in patients with liver cirrhosis and chronic heart failure assessed with the 13C-methacetin breath test, *Ces a Slov Gastroent a Hepatol* 2006, 60 (Suppl1), pages 109
- [7] Málek F, Havrda M, Frühaufová Z, Vránová J: Multidisciplinární přístup v péči o nemocné s kardiorenálním syndromem – první zkušenosti. *Abstrakty, Echodny 2006*, 21.9.-23.9.2006. *Cor Vasa* 2006,48(9): Kardio K 185

